P152

TOTAL ANTIOXIDANT CAPACITY OF SEVERAL SYNTHETIC PHENOLS BY A CHEMILUMINESCENCE METHOD

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The following sinthetic compounds: 2,6-di-tert-butylphenol (1), 2,4-di-tert-butylphenol N,N-dimethyl-3,5-di-*tert*-butyl-4-hydroxybenzylamine 3.5-di-tert-butyl-4-(2),(3): hydroxybenzyl hydrazine (4); 1-(3',5'-di-*tert*-butyl-4'-hydroxybenzyl)-3,5-dimethylpyrazole (5); 1-(3',5'-di-tert-butyl-4'-hydroxybenzyl)-3-methylpyrazol-5-one (6) and 1-(3',5'-di-tertbutyl-4'hydroxybenzyl)-3,5-diphenylpyrazole (7) were studied for their total antioxidant capacity (TAC). All compounds are sterically hindered phenols. Of these the first two are commercially available and (3) - (7) were synthesized in laboratory [1]. TAC of the compounds dissolved in ethanol: EDTA 2x10⁻⁴M=80% (v/v) solution was studied by means of an "in batch" chemiluminometric method based on luminol-Co(II)/EDTA-H₂O₂ system [2]. Calibration curves for trolox (5x10⁻⁶ and 10⁻³M) and gallic acid (10⁻⁵ and 2x10⁻³M) were drawn and results were expressed as % Trolox equivalents (TE) and % gallic acid equivalents (GAE). Of the seven compounds analyzed, (5) has the highest antioxidant capacity (% TE = 27.3 ± 2.4 and % GAE = 63.7 ± 5.5) followed by (2) (% TE = 2.31 ± 0.13 and % GAE = 6.10 ± 0.030) and the other compounds with smaller values (< 0.5 % TE and <1.7% GAE) of TAC.

References:

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